| Prior Learning | | | | |
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| Lesson Number | AQA Spec | Title | Content | Assessment/ Homework |
| 1 | 4.5.1.1 4.5.1.2 4.5.1.3 | Key concept: Endothermic and exothermic reaction | Explore the temperature changes produced by chemical reactions. Consider how reactions are used to heat or cool their surrounding. Investigate how these temperature changes can be controlled. | |
| 2. | | Required practical: investigate the variables that affect temperature changes in reacting solutions, such as acids plus metals, acids plus carbonates, neutralisations, displacement of metals | Devise a hypothesis. Devise an investigation to test your hypothesis. Decide whether the evidence supports your hypothesis. Recap all general equations from C4 | Assessment: Teacher marked Exampro questions linked to required pracitcal |
| 3 | 4.5.1.2 | Reaction profiles | Use diagrams to show the energy changes during reactions. Show the difference between exothermic and endothermic reactions using energy profiles. Find out why many reactions start only when energy or a catalyst is added. | |
| 4 Higher tier only | 4.4.1.3 | Energy change of reactions | Identify the bonds broken and formed during a chemical reaction. Consider why some reactions are exothermic and others are endothermic. Use bond energies to calculate overall energy changes. | Assessment: 25-mark quiz Self-assessed |

| 5 TRIPLE ONLY | 4.5.2.1 | Cells and batteries | Make simple cells and measure their voltages. Consider the importance of cells and batteries. Find out how larger voltages can be produced | |
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| 6 TRIPLE ONLY | 4.5.2.2 | Fuel cells | Find out how fuel cells work. Compare and contrast the uses of hydrogen fuel cells, batteries and rechargeable cells. Learn what reactions take place inside hydrogen fuel cells. | |
| 7 | 4.3.1.4 | Chemical measurements and uncertainty | Explore ideas about the accuracy of measurements. Consider how closely measurements reflect true values. Consider uncertainties of equipment and measurements Calculate percentage uncertainty | |
| | | | End of Unit test | |
| Where we will use these ideas again | | | | |